

## Microstructure and varistor properties of ZnO–V<sub>2</sub>O<sub>5</sub>–MnO<sub>2</sub>-based ceramics

Choon-Woo Nahm

Received: 19 April 2007 / Accepted: 18 June 2007 / Published online: 6 July 2007  
© Springer Science+Business Media, LLC 2007

Recent developments in electronic design have tended toward smaller and higher density packaging of circuitry. This results in a greater susceptibility to surges. Once attacked by surge, electronic circuits can be destroyed in as short as 0.1  $\mu$ s. ZnO doped with several different metal oxides are semiconducting ceramics possessing varistor behaviors, which exhibit abruptly increasing current in accordance with increasing voltage. This non-ohmicity of current–voltage properties is due to the presence of a double Schottky barrier (DSB) formed at active grain boundaries containing many trap states. Owing to highly non-ohmicity, these ceramic devices are widely used in the field of overvoltage protection systems [1, 2]. ZnO ceramics cannot exhibit a varistor behavior without adding heavy elements with large ionic radii such as Bi, Pr, Ba, etc. Commercial Bi<sub>2</sub>O<sub>3</sub>- and Pr<sub>6</sub>O<sub>11</sub>-based ZnO varistor ceramics cannot be co-fired with a silver inner-electrode (m.p. 961 °C) in multilayered chip components because of the relatively high sintering temperature above 1000 °C [3, 4]. Therefore, new varistor ceramics are required in order to use a silver inner-electrode. Among the various ceramics, one candidate is the binary ZnO–V<sub>2</sub>O<sub>5</sub> system [5–9]. This system can be sintered at relatively low temperature in the vicinity of about 900 °C. This is important for multilayer chip component applications, because it can be co-sintered with a silver inner-electrode without using expensive palladium or platinum metals.

To develop varistor ceramics of high performance, it is very important to comprehend the influences of additives on varistor properties. MnO<sub>2</sub> is often added to Bi<sub>2</sub>O<sub>3</sub>-doped

ZnO varistors to improve the varistor properties [10, 11]. In this report, the influence of MnO<sub>2</sub> on the microstructure and varistor properties of ternary ZnO–V<sub>2</sub>O<sub>5</sub>–MnO<sub>2</sub> (ZVM)-based ceramics was examined.

Reagent-grade raw materials were prepared for ZnO non-ohmic ceramics with a ternary composition, such as (99.5– $x$ ) mol% ZnO + 0.5 mol% V<sub>2</sub>O<sub>5</sub> +  $x$  mol% MnO<sub>2</sub> ( $x = 0.0, 0.25, 0.5, 1.0, 2.0$ ). Raw materials were mixed by ball-milling with zirconia balls and acetone in a polypropylene bottle for 24 h. The powder was pressed into discs 10 mm diameter and 2 mm thickness at a pressure of 80 MPa. The discs were sintered at 900 °C for 3 h. The final samples were 8 mm in diameter and 1.0 mm in thickness. Silver paste was coated on both faces of the samples and the electrode was formed by heating at 600 °C for 10 min. The area of electrodes was approximately 0.196 cm<sup>2</sup>.

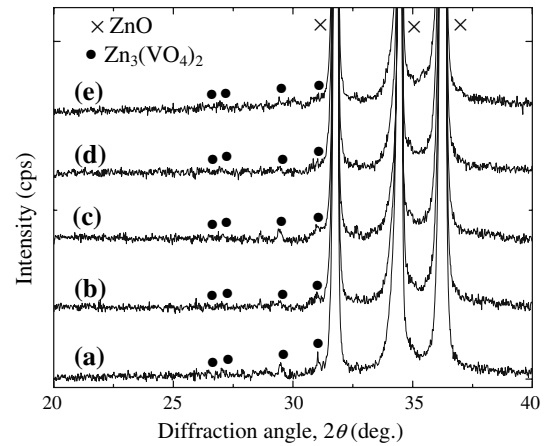
The surface microstructure was examined by scanning electron microscopy (SEM, Model S2400, Hitachi, Japan). The average grain size ( $d$ ) was determined by the linear intercept method [12]. The crystalline phases were identified by powder X-ray diffraction (XRD, Model D/max 2100, Rigaku, Japan) with CuK $\alpha$  radiation. The sintered density ( $\rho$ ) was measured by the Archimedes method. The voltage–current (V–I) characteristics were measured using an I–V source (Keithley 237). The breakdown voltage ( $V_B$ ) was measured at 1.0 mA/cm<sup>2</sup> and the leakage current ( $I_L$ ) was measured at 80–% of the breakdown voltage. In addition, the non-ohmic coefficient ( $\alpha$ ) is defined by the empirical law,  $J = K \cdot E^\alpha$ , where  $J$  is the current density,  $E$  is the applied electric field, and  $K$  is a constant.  $\alpha$  was determined in the current density range of 1.0 mA/cm<sup>2</sup> to 10 mA/cm<sup>2</sup>, where  $\alpha = 1/(\log E_2 - \log E_1)$ , and  $E_1$  and  $E_2$  are the electric field corresponding to 1.0 mA/cm<sup>2</sup> and 10 mA/cm<sup>2</sup>, respectively. Five samples for non-ohmic resistors (sintered at the same

C.-W. Nahm (✉)  
Department of Electrical Engineering, Dongeui University,  
Busan 614-714, Korea  
e-mail: cwnahm@deu.ac.kr

time) were used for all electrical measurements and their average value is presented.

Figure 1 shows SEM micrographs of the ZVM-based ceramics containing different amounts of  $MnO_2$ . The grain structure is very heterogeneously distributed throughout the samples. The undoped samples showed abnormal grain growth of ZnO. The non-uniformity of grain size was significantly reduced with an increase of  $MnO_2$  dopant levels. With increasing  $MnO_2$  dopant levels, the average grain size decreased from 16.2 to 5.2  $\mu m$ . Therefore, the incorporation of  $MnO_2$  effectively reduced abnormal grain growth. At the same time, doping with  $MnO_2$  did not significantly modify the densification process. The sintered density was in the range of 94.6% to 95.7% of the theoretical density (TD) (pure ZnO, TD = 5.78  $g/cm^3$ ). The XRD patterns of the ZVM-based ceramics are shown in Fig. 2. These patterns revealed the presence of  $Zn_3(VO_4)_2$  as a secondary phase, in addition to primary phase of hexagonal ZnO. No secondary phase related to  $MnO_2$  was detected. The detailed microstructure parameters are summarized in Table 1.

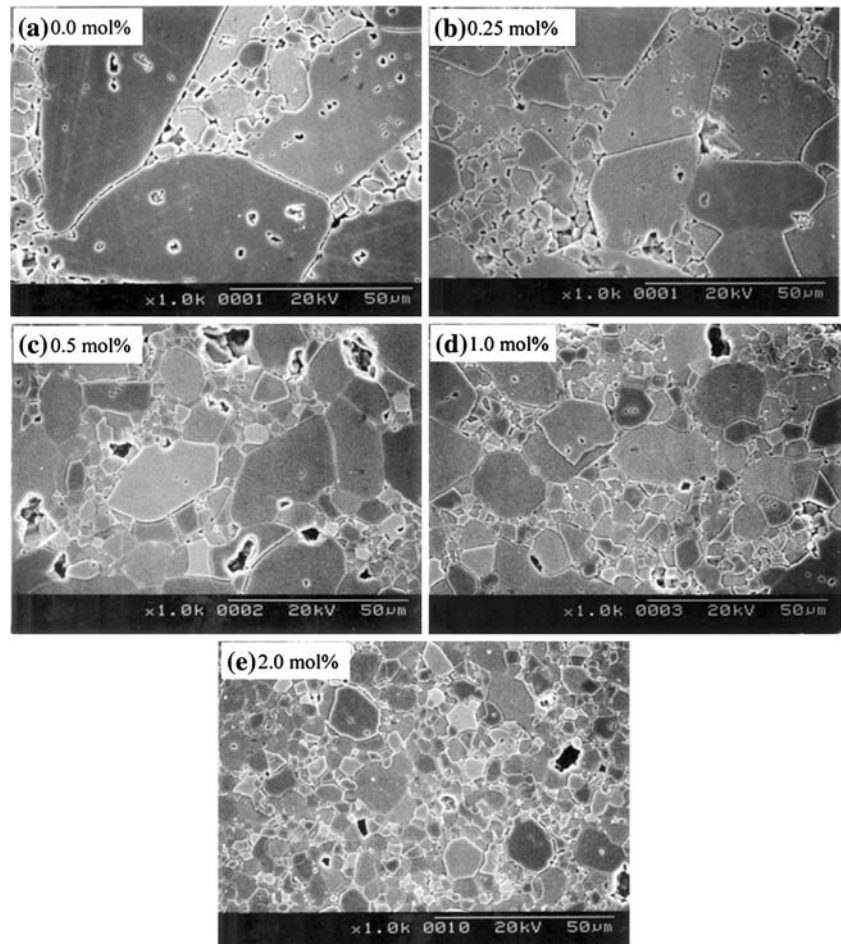
Figure 3 shows the electric field–current density (E–J) characteristics of the ZVM-based ceramics for different



**Fig. 2** XRD patterns of the ZVM-based ceramics for different amounts of  $MnO_2$ : (a) 0.0 mol%, (b) 0.25 mol%, (c) 0.5 mol%, (d) 1.0 mol%, and (e) 2.0 mol%

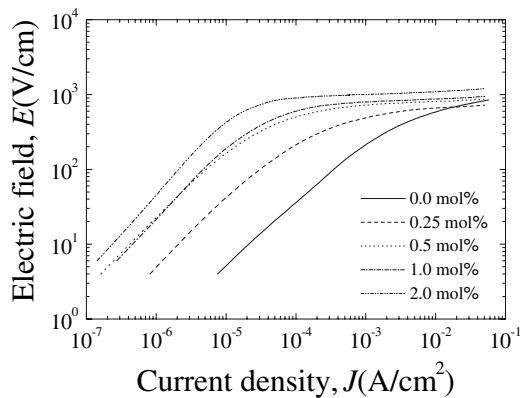
amounts of  $MnO_2$ . The varistor properties are characterized by non-ohmicity in the E–J characteristics. The curves show the conduction characteristics divide into two regions: an ohmic region before breakdown and a non-ohmic region after breakdown. The sharper the knee of the

**Fig. 1** SEM micrographs of the ZVM-based ceramics for different amounts of  $MnO_2$



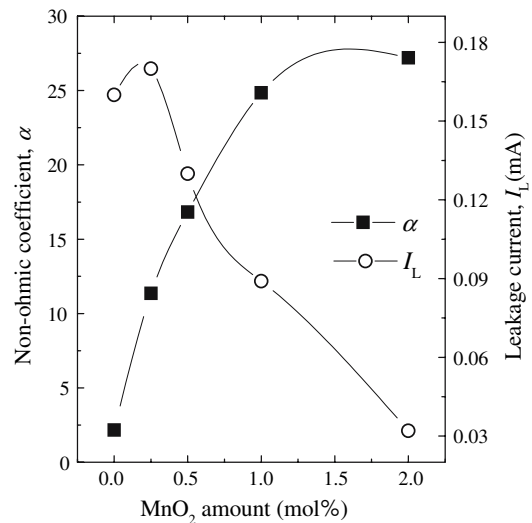
**Table 1** Microstructure and V–I characteristic parameters of the ZVM-based ceramics for different amounts of MnO<sub>2</sub>

MnO <sub>2</sub> amount (mol%)	d (μm)	ρ (g/cm <sup>3</sup> )	V <sub>B</sub> (V/mm)	V <sub>b</sub> (V)	α	I <sub>L</sub> (μA)
0.0	16.1	5.53	17.5	0.28	2.2	0.16 mA
0.25	12.2	5.50	53.5	0.65	11.4	0.17 mA
0.5	8.4	5.47	69.9	0.58	16.8	0.13 mA
1.0	7.9	5.48	78.1	0.62	24.9	89.4
2.0	5.2	5.51	99.2	0.52	27.2	32.9

**Fig. 3** E–J characteristics of the ZVM-based ceramics for different amounts of MnO<sub>2</sub>

curves between the two regions, the better the non-ohmic properties. The undoped samples showed very poor non-ohmic properties. On adding more MnO<sub>2</sub>, the knee gradually becomes more pronounced and the varistor properties are enhanced. Therefore, the incorporation of MnO<sub>2</sub> seems to remarkably enhance varistor properties. The breakdown voltage ( $V_B$ ) increased from 17.5 to 99.2 V/mm with increase in MnO<sub>2</sub> concentration. The increase of  $V_B$  with increasing MnO<sub>2</sub> concentration can be explained by the increase in the number of grain boundaries owing to the reduction in the average ZnO grain size. The breakdown voltage per grain boundary ( $V_b$ ) for the ZVM-based ceramics was in the range of 0.5–0.6 V, which exhibited a much lower  $V_b$  value than the general value of 2–3 V for Bi<sub>2</sub>O<sub>3</sub> and Pr<sub>6</sub>O<sub>11</sub>-based ZnO varistors. This suggests that the ZVM-based ceramics can be used in the manufacture of low-voltage varistors.

Figure 4 shows the variation of the non-ohmic coefficient ( $\alpha$ ) and the leakage current ( $I_L$ ) of the ZVM-based ceramics as a function of MnO<sub>2</sub> concentration. The  $\alpha$  and  $I_L$  values are derived from the E–J curves shown in Fig. 3. The  $\alpha$ -value of undoped samples was only 2.2, whereas the  $\alpha$  value of MnO<sub>2</sub>-doped samples significantly increased from 11.4 to 27.2. In particular, when the MnO<sub>2</sub> level is more than 1.0 mol%, the samples exhibited relatively good varistor properties. The maximum non-ohmic coefficient was 27.2, which was obtained from the addition of 2.0 mol% MnO<sub>2</sub>. The  $I_L$  value was not affected much by

**Fig. 4** Non-ohmic coefficient and leakage current of the ZVM-based ceramics as a function of MnO<sub>2</sub> concentration

MnO<sub>2</sub> concentration less than 0.5 mol%. However, the  $I_L$  value decreased to 32.9 μA with the addition of 2.0 mol% MnO<sub>2</sub>. It resulted in the conclusion that doping the binary ZnO–V<sub>2</sub>O<sub>5</sub>-based ceramics with MnO<sub>2</sub> remarkably improves the non-ohmic properties, resulting in a higher non-ohmic coefficient and a smaller leakage current. The detailed V–I characteristic parameters are summarized in Table 1.

In summary, for all samples, the microstructure of the ZnO–V<sub>2</sub>O<sub>5</sub>–MnO<sub>2</sub>-based ceramics consisted of ZnO grains and Zn<sub>3</sub>(VO<sub>4</sub>)<sub>2</sub> as a secondary phase. The addition of MnO<sub>2</sub> to the ZnO–V<sub>2</sub>O<sub>5</sub>-based ceramics was found to reduce abnormal grain growth of ZnO. The varistor properties were improved with the increase of MnO<sub>2</sub> concentration and a maximum non-ohmic coefficient ( $\alpha = 27.2$ ) was obtained for the samples containing 2.0 mol% MnO<sub>2</sub>. Thus the ZnO–V<sub>2</sub>O<sub>5</sub>–MnO<sub>2</sub>-based ceramics is a potential candidate for chip varistors with a silver inner-electrode.

## References

- Levinson LM, Pilipp HR (1986) Am Ceram Soc Bull 65:639
- Gupta TK (1990) J Am Ceram Soc 73:1817

3. Nahm C-W (2001) Mater Lett 47:182
4. Nahm C-W (2002) J Mater Sci Lett 21:201
5. Tsai J-K, Wu T-B (1994) J Appl Phys 76:4817
6. Tsai J-K, Wu T-B (1996) Mater Lett 26:199
7. Kuo CT, Chen CS Lin I-N (1998) J Am Ceram Soc 81:2942
8. Hng H-H, Knowles KM (1999) J Eur Ceram Soc 19:721
9. Hng H-H, Halim L (2003) Mater Lett 57:1411
10. Pike GS, Kurtz SR, Gourley PL, Philipp HR, Levinson LM (1985) J Appl Phys 57:5512
11. Greuter F, latter G, Rossineli M, Stucki F (1989) Am Ceram Soc Westerville OH 3:31
12. Wurst JC, Nelson JA (1972) J Am Ceram Soc 55:109